



Contents lists available at ScienceDirect

Journal of King Saud University – Science

journal homepage: www.sciencedirect.com



Original article

Techno-economic analysis of an integrated electrocoagulation-membrane system in treatment of palm oil mill effluent



Muhammad Syaamil Saad ^a, Ng Chun Joe ^a, Hanifatur Affiq Shuib ^a, Mohd Dzul Hakim Wirzal ^{a,*}, Zulfan Adi Putra ^b, Mohammad Rizwan Khan ^c, Rosa Busquets ^d

^a Chemical Engineering Department, Universiti Teknologi PETRONAS, 32610 Bandar Seri Iskandar, Perak, Malaysia

^b PETRONAS Group Technical Solutions, Project Delivery and Technology, PETRONAS, Kuala Lumpur 50050, Malaysia

^c Chemistry Department, College of Science, King Saud University, Riyadh 11451 Saudi Arabia

^d School of Life Sciences, Pharmacy and Chemistry, Kingston University London, Penrhyn Road, Kingston upon-Thames, Surrey KT1 2EE, United Kingdom

ARTICLE INFO

Article history:

Received 6 January 2022

Revised 3 February 2022

Accepted 31 March 2022

Available online 6 April 2022

Keywords:

Palm oil mill effluent
POME treatment
Electrocoagulation
Membrane
Techno-economic analysis

ABSTRACT

Palm oil mill effluent (POME) is an oily wastewater generated by palm oil processing mills. Most palm oil factories and refineries have their own treatment system such as anaerobic digestion system and electrocoagulation (EC). In this study, a new method, integrated electrocoagulation (EC)-membrane system is introduced to treat the POME. The treatment efficiency and economic feasibility of the developed integrated EC-membrane is also evaluated by comparing with EC standalone system. This study highlights that the removal of POME by EC has increased with extended time and increasing voltage. Besides, the integrated EC-membrane shows better removal efficiency compared to the EC system. For model development, both standard least squares model for EC and integrated membrane-EC systems created by using JMP software showed a good fit and able to be used for further efficiency calculation. The economic analysis result shows that integrated EC-membrane method has a higher capital cost with lower in the operating cost per year than EC method. 75% color removal efficiency is achievable by integrated EC-membrane system at voltage 1 V and 2 V in which EC method requires voltage of 4 V and higher. Notably, the total cost of integrated EC-membrane system at the first year of implementation is lower than the EC system and in 15 years period has a 21.24% lesser cost compared to EC system.

© 2022 The Author(s). Published by Elsevier B.V. on behalf of King Saud University. This is an open access article under the CC BY-NC-ND license (<http://creativecommons.org/licenses/by-nc-nd/4.0/>).

1. Introduction

Palm oil mill effluent (POME) is the waste product generated by the palm oil processing mills that contains of different suspended materials and dissolved solids. It appears as a thick brown saturated liquid waste with unlikable odor with soluble solids and materials in it that has an enormous impact to the environment (Ali Huddin, Irvan, Mohd Nordin, & Arezoo Fereidonian, 2012; Saad et al., 2020b; Tan & Lim, 2019). POME has potential to pollute the environment and several treatment processes has been developed to treat POME before it is discharged out to the environment.

The existing treatment processes of POME are anaerobic digestion, EC, membrane, and adsorption. Anaerobic digestion process is economically feasible as the method needs a low capital and operating cost by using covered lagoon as the digester type. (Choong, Chou, Norli, & Reviews, 2018; Medri & Costa, 2003; Sani et al., 2020). However, anaerobic digestions require large land area, long hydraulic retention time (HRT) and produces smelly odor (Lee et al., xxxx; Lee & San Lee, 2020; Wastewater Technology Fact Sheet Anaerobic Lagoons, 2002; Zainal, 2018).

Recent studies have shown that EC system is an excellent alternative to palm oil-based wastewater treatment (Bashir, Mau Han, Jun Wei, Choon Aun, & Abu Amr, 2016; Chairunnisa, Arifin, Sofyan, Lubis, & Darmadi, 2018; Nasrullah et al., 2017). EC is an electrochemical method that uses aluminum (Al) or iron (Fe) electrodes powered by direct current (DC) to treat waste effluents. The metal ions generated at the anode will undergo hydrolysis or oxidation process in the water and produced coagulant hydroxide precipitates which will then be denser than water and able to be removed by settling process and adsorption (Hakizimana et al., 2017). Meanwhile the cathode will produce hydroxyl ions that

* Corresponding author.

E-mail address: mdzulhakim.wirzal@utp.edu.my (M.D.H. Wirzal).

Peer review under responsibility of King Saud University.



Production and hosting by Elsevier

forms hydrogen gas and oxygen by reduction process. The hydrogen and oxygen gases produced will result in floating effect, which causes the pollutants to rise to the surface and form a stable floc layer (Syaichurrozi et al., 2020; Wirzal et al., 2020; Wirzal et al., 2013). Usually, Al and Fe is chosen as the electrodes as they have advantages such as low price, non-toxic and have a better removal efficiency of pollutants. Besides, EC system on POME shows good result in reducing the turbidity of the effluent, increasing the pH value close to neutral, and greatly reducing the COD and BOD level (Barrera-Díaz, Frontana-Uribe, & Bilyeu, 2014; Bashir, Lim, Abu Amr, Wong, & Sim, 2019; Nasrullah et al., 2017).

Membrane technology is an effective and economical separation method used widely in industrial wastewater treatment (Abd Halim et al., 2019; Abdurakhman et al., 2018a; Zahari et al., 2018; Zhang, Yao, Maleki, Liao, & Lin, 2019). The membrane served as a barrier to block and separate the materials with particular affinity through pore management. A few of its advantages are simple and easy operation, low energy consumption and no use of chemicals (Abdurakhman et al., 2018b; Chen et al., 2018; Ma, Ping, & Dong, 2017). The membrane used in this research is nylon 6,6 nanofiber membrane (NFM). Nylon 6,6 has shown better advantages than other membranes like higher fouling resistant and clean water permeability (Abd Halim et al., 2019; Bilad et al., 2018; Jasni et al., 2017; Mat Nawi et al., 2020). The membrane separation treatment was reported to have greatly reduce the turbidity, COD and BOD amount by using ultrafiltration (UF) and reverse osmosis (RO) (Latif Ahmad, Ismail, & Bhatia, 2003). However, membrane technology has a major drawback, which is membrane fouling that reduce the efficiency of separation, especially membranes with smaller pores such as nanofiltration, UF and RO (Eliseus et al., 2018). The fouling situation can be delayed by having a pre-treatment such as EC or coagulation which aid filtration process by agglomerating small particles into larger masses and make it easier to be eliminated (Abdulsalam, Che Man, Isma Idris, Faezah Yunos, & Zainal Abidin, 2018; Elma, Rahma, Pratiwi, & Rampun, 2020; Xing et al., 2019). In addition to that, the economic aspects of newly developed systems are rarely reported for wastewater treatment methods. Many previous studies managed to report the performance of the treatment systems that were developed. Yet, economic aspects of it were rarely studied or mentioned.

To tackle the issues mentioned, a new system, integrated EC-membrane system was proposed which will provide higher removal efficiency with lower energy consumption. The proposed method started with EC system that coagulate the particles in the POME and transfer the wastewater along with the suspended solid through the membrane to filter out the solid particles. Furthermore, economical aspects of the reported results by Muhammad Syaamil Saad et al. (2020) were emphasized in this study. Thus, the aim of this paper is to perform modelling on the obtained experimental data and subsequently evaluate the economic analysis of the proposed methods. Parameters such as voltage and time between integrated membrane-EC system and EC standalone method was assessed to determine its effectiveness in color removal as well as maintain a low operating cost if it is to be scaled up to industrial size.

2. Materials and method

2.1. Chemicals and apparatus

Chemicals used in this research were formic acid (98–100%) (Merck, USA), acetic acid (Merck, USA), nylon 6,6 pellets (Sigma, USA) and sodium chloride (Merck, USA). Fe plates used for electrodes were procured from local store. Apparatus involved in performing integrated EC-membrane treatment of POME includes

150 mm diameter filter paper (Whatman, UK), DC power supply (Zhaoxin, Model: RXN303D, China), Magnetic stirrer (IKA, Model: C-MAG HS 7, Germany), peristaltic pump (Chuang Rui, Model: BT600M, China), 5 ml syringe (TERUMO, Japan), needle (TERUMO, Japan), syringe pump (Nanolab, Model: SP20, Malaysia), high voltage power supply (Nanolab, Model: PS35-PV, Malaysia), and a rotating drum controller (Nanolab, Model: ESD 30, Malaysia).

2.2. EC setup

The EC setup was executed and adopted from (Muhammad Syaamil Saad et al., 2020) as this study is the continuation of the previous research. Fe plates with dimension of 6×4 cm were used as electrodes. In summary, the voltage applied for EC treatment of POME were 2 V (1.50 mA/cm^2), 4 V (4.75 mA/cm^2), 6 V (5.00 mA/cm^2), 8 V (6.50 mA/cm^2) and 10 V (110.00 mA/cm^2). A 400 ml of pre-treated of POME samples were placed in a 500 ml beaker and stirred continuously for 60 min during EC with Fe electrodes. Every 15 min, 30 ml of POME were filtered and collected for the determination of colour removal using UV-Vis spectrophotometry.

2.3. Nylon 6,6 membrane fabrication

Nylon 6,6 membrane fabrication method was also adopted from Muhammad Syaamil Saad et al. (2020) and Abd Halim et al. (2019). Nylon 6,6 pellet was added to the mixture of formic acid and acetic acid (14:86) of polymer to solvent. Membrane solutions were stirred for 24 h under temperature of 70 °C to allow homogeneous mixture. Membrane was then fabricated by electrospinning method using 20 KV of voltage supplied, membrane solution injection flowrate of 0.4 ml/h and spinning collector speed of 500 rpm for 37 h.

2.4. Integrated EC-membrane setup

The membrane sheet was cut and put between two rubber gaskets which were connected via rubber pipe from the POME. The edges of the rubber frame were enveloped with polytetrafluoroethylene (PTFE) tape to prevent leakage at the side and was clipped both left and right sides to hold the two frames tighter. The feed and retentate were dipped inside the beaker containing POME solution while the permeate was channeled into a measuring cylinder for collection. Before pump was turned on, initial sample of POME was collected for the analysis and comparison with permeate. Then, the permeate was collected every 15 min and volume was measured to calculate the membrane permeability. Once the permeate reached steady state (constant volume for 3 consecutive sampling), DC power supply was turned on. The samples collections were done at permeate side of membrane and were taken every 15 min for an hour with 1 V (0.05 mA/cm^2) and 2 V (1.50 mA/cm^2) applied to the electrode. Lower voltage (1 V and 2 V) was specifically chosen to reduce the energy consumption and hence the cost. With integration of membrane, lower voltage was redeemed sufficient to maintain a high removal efficiency of color of POME. Finally, the permeate was collected for analysis and pump was turned off. The integrated EC-membrane setup is illustrated as Fig. 1.

2.5. Techno-economic analysis

2.5.1. Data collection

Data required for modelling were collected from articles, journals, experiment and thesis. The parameters that were collected are the UV-Vis and COD reduction result based on the independent variables of types of electrodes, voltage and time. The experiments were repeated three times for validation of the results.

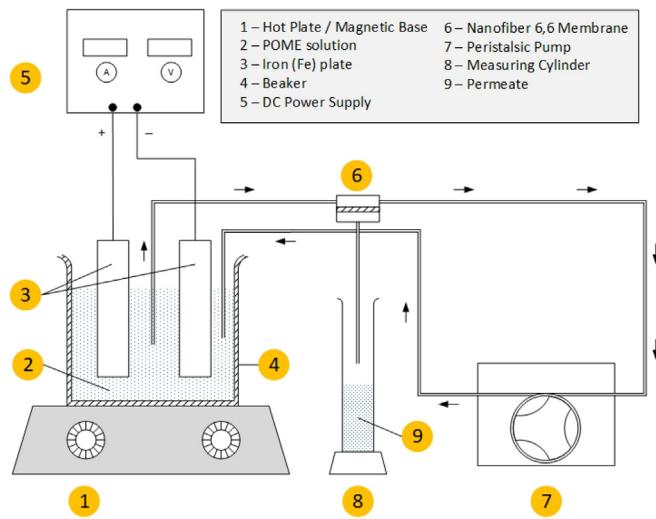


Fig. 1. Integrated electrocoagulation (EC)-membrane setup of POME with 0.1 bar pressure and 27 °C using nylon 6,6 nanofiber as membrane.

2.5.2. Statistical data analysis

After the data collection, the data was then analysed by calculating the percentage of removal by each method. Furthermore, the data have then been developed into a model by using the JMP Pro 13.0 software for EC and integrated EC-membrane method. Graphs were then plotted by the removal efficiency and the manipulated variables such as voltage and time.

2.5.3. Modelling and validation

Modelling of each method were done to evaluate the significant of the manipulated variables to the process. Regression analyses were used to predict the modelling of both EC and EC-membrane system. Standard Least Square Model is an approach of a regression analysis by minimizing the sum of squares of residuals to create a fitting model. The modelling method optimize the application of data fitting as the fitting of the model is based on the best fit line of data and helps to estimate the relationship between a dependent variable towards one or a few sets of independent variables. The method often used in linear regression model is ordinary least square estimate. This method generates estimators (constant variable) for each of the independent variables in the model. The graph of the removal efficiency against the predicted value was plotted to determine the correlation between each variable while determining the value of R-squared (RSq), root mean square error (RMSE) and the P value.

2.5.4. Economic analysis

Capital expenditure (CAPEX) and operating expenses (OPEX) comparison was made between each model to determine the best option that exhibits the minimum cost to run a treatment plant using Microsoft Excel and OriginPro 2016 software which is a data analysis and graphing software. According to Smith (2005a), CAPEX can be estimated using fixed factors as reported in Table 1 for economic analysis of a plant. Hence, the values were adopted into the calculation of plant erection in this study.

3. Result and discussion

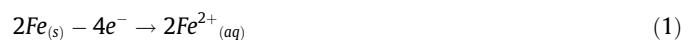
3.1. Comparison of method

The removal efficiency of the proposed method, integrated EC-membrane system was compare with EC standalone system

Table 1
Factors Included for Capital Cost Calculation.

Symbol	Item	Factor
f_1	Equipment Erection (Installation)	0.4
f_2	Piping	0.7
f_3	Instrumentation and control system	0.2
f_4	Electrical System	0.1
Physical Plant Cost Factor		1.4
f_5	Design and Engineering Fees	0.3
f_6	Contractor's Fees	0.05
f_7	Contingency Cost	0.1
Fixed Capital Factor		0.45

(Muhammad Syaamil Saad et al., 2020). Both system were previous related research from the same author. The proposed integrated system is a combination of EC method followed by a microfiltration process by using nylon 6,6 nanofiber membrane as a filtration unit. Fig. 2 shows the removal efficiency for EC and integrated EC-membrane methods. From the results, integrated system exhibits excellent removal efficiency even at low voltages (1 V and 2 V). Color removal efficiency of integrated system for 1 V at 15 min was 72.63%, then increased to 74.64%, 75.00% and 75.09% for every next 15 min until 60 min. The same trend was observed for integrated system at 2 V. The first 15 min shows the lowest color removal efficiency which is 74.30%, then dropped to 75.52%, 76.79% and 76.99% with every 15 min intervals. This increasing trend is often exhibits by an EC treatment. The trend observed by EC standalone systems are proportional to the increase in voltage and time in which aligns with most of treatment done by EC (Aris, Ibrahim, Arifin, & Hawari, 2017; Ibrahim, Aris, Arifin, Hawari, & Hanafiah, 2018; Saad et al., 2020a; Yavuz & O gutveren, 2018). This is the norm when dealing with EC as the higher the voltage applied, the higher the rate of anode dissolution hence release more coagulant agent (Fe^{2+}). Formation of bigger colloids can be concluded as more sedimentation can be observed and improved decolorization of treated wastewater as voltage and time increase. In addition, at low voltage (2 V), EC standalone method removal efficiency of color is relatively low when compared to integrated system with minimum of 13.40% and maximum color removal of 18.90%. At 4 V, color removal efficiency increased from 46.40% to 63.20%, 72.7% and 76.20% for increment every 15 min, which is comparable to 1 V and 2 V in integrated system. As the voltage was applied to the electrodes and the reaction time was prolonged, more metal atoms will turn into ions. In this study, more Fe^{3+} and Fe^{2+} ions were formed on the anode surface as in reaction in Equation 1–3 as reported (Chafi, Gourich, Essadki, Vial, & Fabregat, 2011; Irki, Ghernaout, Naceur, Alghamdi, & Aichouni, 2018).



From Fig. 2, integrated EC-membrane system has shown interesting results where the removal efficiency of color for 1 V and 2 V showed excellent color removal rate even at a lower voltage compared to normal EC process. This phenomenon may be explained by formation of colloids that flocs together during EC, making it larger in size (Gong et al., 2017). Integrated EC-membrane system has shown interesting results where the removal efficiency of color for 1 V and 2 V shows high color removal even at low voltage as compared to normal EC process. This phenomenon may be explained by the formation of colloids

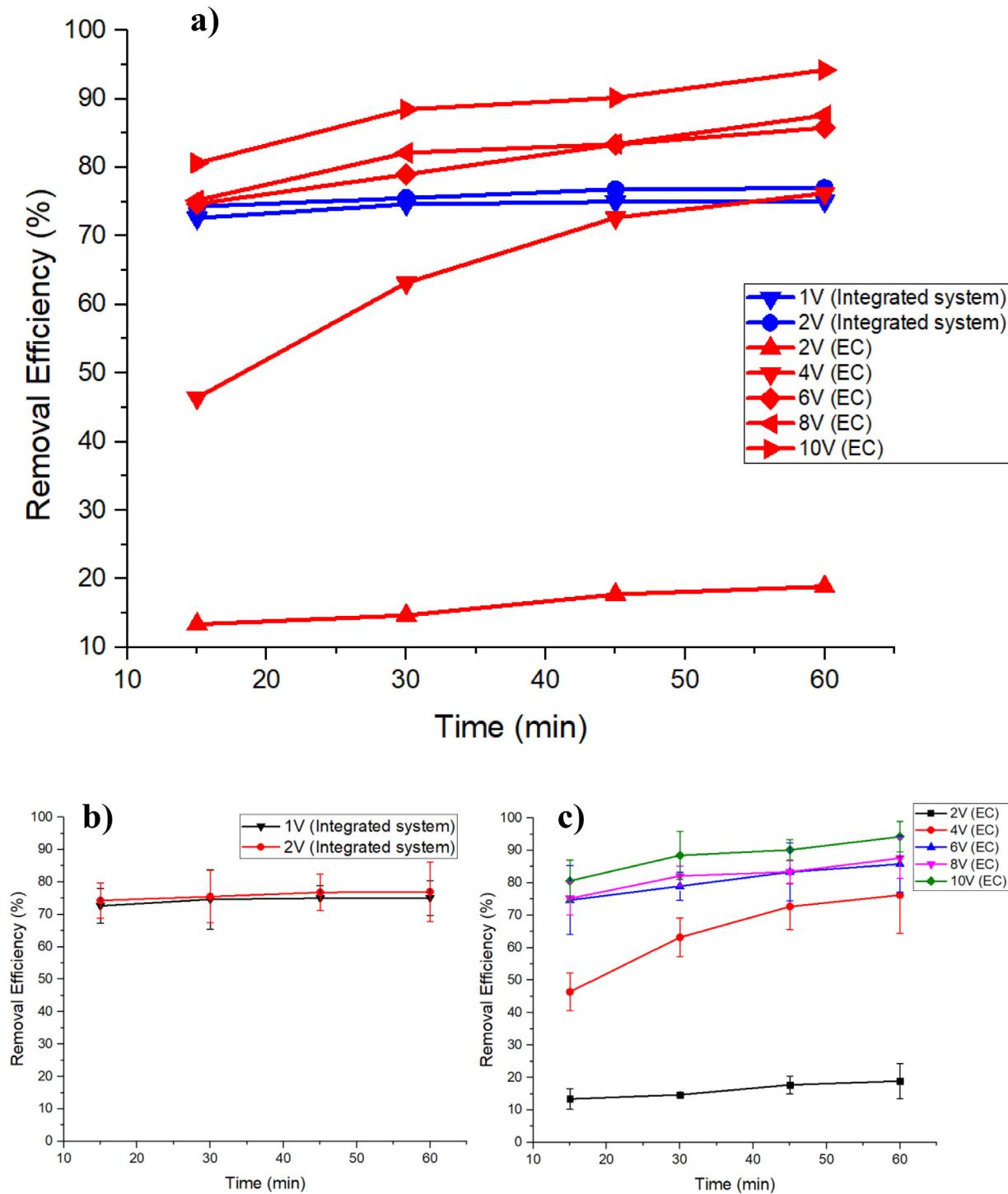


Fig. 2. **a)** Removal efficiency of color at 1 V and 2 V (Integrated EC-membrane) and 2 V, 4 V, 6 V, 8 V, 10 V (EC standalone) of POME, **b)** Removal efficiency of integrated system with error bars and standard deviations (1 V and 2 V), and **c)** Removal efficiency of EC system with error bars and standard deviations (2 V, 4 V, 6 V, 8 V and 10 V).

that flocs together during electrocoagulation, making it larger (Gong et al., 2017). One of the reasons why the integration of nano-fiber was crucial was to filter the colloids from wastewater and allowing a much clearer solution to pass through its pores (Saad et al., 2020a). For 1 V, the low reaction of coagulants with pollutants in wastewater caused by the low anode dissolution resulting in less formation of bigger colloids, hence pass-through membrane pores. This explains the slightly lower removal efficiency of color as compared to 2 V in integrated system. Thus, to counter the prob-

lem, the voltage was increased. This was well exhibited by the integrated system at 2 V where the removal efficiency of colour at the end of treatment was seen slightly improved compared to 1 V. With only microfiltration treatment, the removal colour from POME was only 13.6%. It was proven that with integration of EC at low voltage could further increase the removal efficiency of colour in POME. In alignment with the scope of this study, lower voltage (1 V and 2 V) were investigated on how it affects integrated EC-membrane system despite the potential for higher voltage to

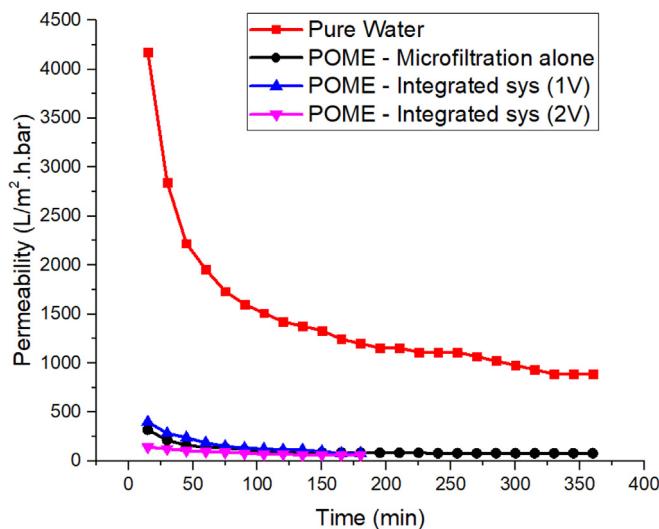


Fig. 3. Permeability comparison between pure water and POME with microfiltration standalone and integrated system at 1 V and 2 V (Muhammad Syaamil Saad et al., 2020).

reduce more colour. This was because at higher voltage (> 2 V), the system would incur additional costs. Therefore, process modeling and techno-economic analysis were done to evaluate which system will benefit the most with the high removal efficiency of color. This is why the integration of nanofiber is crucial as the membrane will filter the colloids from wastewater and allowing a much clearer solution to pass through its pores (Saad et al., 2020a).

Membrane permeability in the integrated system has shown a slight increment in the early treatment of POME using 1 V as shown in Fig. 3. However, the steady-state of permeate flowing through the membrane was observed to be approximately the same as microfiltration method alone. The coagulation of the particles during EC process succeeded in reducing the fouling in the early stages of microfiltration and at a lower voltage (1 V). Despite that, the same results were not observed in 2 V. This phenomenon can be explained by increment of voltage during EC that resulted in a higher coagulation rate of the particles. As the voltage increases, fouling immediately occur in the early stage of microfiltration and reduces the permeate flow through the membranes. The steady-state of the integrated system at 2 V was also seen to be achieved at a faster rate as compared to 1 V and microfiltration alone.

3.2. Development of model

Based on the experimental data from Saad et al. (2020a) and data from this research, the models for removal efficiency of EC were constructed which were linear, quadratic and cubic equations. In this study, the manipulated variables considered were voltage (V) and time (T). Therefore, the linear equations would

consist of V and T, quadratic equations would have V^2 , V, T^2 , T and cubic equations would be having variables of V^3 , V^2 , V, T^3 , T^2 , T.

Table 2 shows the comparison table of the EC model developed. The result shows quadratic equation model for removal efficiency of EC has the best fit with the RSq adjusted of 0.81 and p-value of 0.0001. Although the cubic equation has a higher RSq adjusted of 0.86 and a lower RMSE of 13.451, it has 3 variables of p-value lower than 0.05 indicates that the variables of the model does not fit well. Hence, quadratic equation model is chosen as the best fit model and is compared with cross-variable model.

The comparison of quadratic equation with quadratic cross variable equation model is further described. The cross variable in this study is combination of voltage and time (VT). The table shows that the quadratic cross variable shows a better fit as it has a lower RMSE of 14.118 and RSq of 0.85 which is higher than the value in quadratic equation model. Therefore, it can be confirmed that the quadratic cross variable model is chosen as the best fit for removal efficiency of EC method. In this model development of EC, no modification is needed as there is no variable with P-value higher 0.05.

All 5 variables showed in the effect summary report shows the p-variable of less than 0.05 indicates all variable are significant and able to affect the removal efficiency. Equation 2 shows the simplified predicted expression of the model produced from JMP software for EC method.

$$\begin{aligned} \text{RemovalEfficiency} = & 11.82 + 6.47V + 1.02T - 1.41(V - 6)^2 \\ & - 0.04(T - 30)^2 + 0.12(T - 30)(V - 6) \end{aligned} \quad (2)$$

The model for color removal efficiency of integrated EC-membrane has also been constructed using experimental data. A total of 3 models constructed consisting of linear, quadratic and cubic equations. **Table 3** shows the comparison of integrated EC-membrane model. The table shows that all three types of equation model are a bad fit. The better fit among the three models created are the cubic equation model as it has the lowest root mean square error (RMSE) of 11.773, highest RSq and RSq adjusted of 0.87 and 0.63 respectively. However, it can be said to be a good fit as according to (Huang & Purvins, 2020), a model should return a p-value of less than 0.05 to be considered as acceptable.

The comparison of modified cubic equation model and modified cubic cross variable model were shown in bottom section of **Table 3**. The table shows that the modified cubic cross variable model that consist of V^3 , T^3 and VVT has a better fit of RSq adjusted of 0.96, among lowest RMSE and lowest p-value of 0.0102. The RMSE shows a promising reduction as well of from 11.773 to 6.215. Thus, the modified cubic cross variable model was the most ideal for integrated EC-membrane system.

The predicted plot from the modelling shows that the R-Square line fits well and has a root mean square (RSq) of 0.96 and p-value of less than 0.0102 indicates that this model shows a good fit. Equation 3 shows the simplified predicted expression for the removal efficiency of integrated EC-membrane method that is being generated by the software from the experimental data.

Table 2
Comparison of EC model.

Type	Variables	RMSE	RSq	RSq (Adjusted)	P-value	Variables P-Value > 0.05
Linear	V, T	23.557	0.62	0.59	0.0001	0
Quadratic	V^2 , V, T^2 , T	15.864	0.84	0.81	0.0001	0
Cubic	V^3 , V^2 , V, T^3 , T^2 , T	13.451	0.9	0.86	0.0001	3
Quadratic	V^2 , V, T^2 , T	15.864	0.84	0.81	0.0001	0
Quadratic Cross Variable	V^2 , V, T^2 , T, VT	14.118	0.88	0.85	0.0001	0

Table 3

Comparison of integrated EC-membrane model.

Type	Variables	RMSE	RSq	RSq (Adjusted)	P-value	Variables P Value > 0.05
Linear	V, T	19.427	0.18	0	0.3988	2
Quadratic	V ² , V, T ² , T	18.639	0.42	0.08	0.3732	4
Cubic	V ³ , V ² , V, T ³ , T ² , T	11.773	0.87	0.63	0.1108	7
Modified Cubic Variable	V, V ² , T, T ² , T ³ , VT	10.901	0.8574	0.69	0.0488	4
Modified Cubic Variable	V, V ² , V ³ , T, T ² , VT	9.978	0.8566	0.74	0.0163	4
Modified Cubic Cross Variable	V, V ² , V ³ , T, T ² , T ³ , VT	10.901	0.8574	0.69	0.0488	5
Modified Cubic Cross Variable	V, V ² , V ³ , T, T ² , T ³ , VT, VTT	11.774	0.8669	0.63	0.1108	7
Modified Cubic Cross Variable	V, V ² , V ³ , T, T ² , T ³ , VT, VVT	6.215	0.9629	0.96	0.0102	4
Modified Cubic Cross Variable	V, V ² , V ³ , T, T ² , T ³ , VT, VVT, VTT	6.187	0.9724	0.89	0.0286	6

$$\begin{aligned} RemovalEfficiency = & 120.05 - 5.62V - 2.13T + (V - 4.33) \\ & \cdot ((V - 4.33) \cdot 2.60) + (V - 4.33) \\ & \cdot ((T - 37.5) \cdot -0.10) + (T - 37.5) \\ & \cdot ((T - 37.5) \cdot 0.002) + (V - 4.33) \\ & \cdot ((V - 4.33) \cdot ((V - 4.33) \cdot 0)) \\ & + (V - 4.33) \\ & \cdot ((V - 4.33) \cdot ((T - 37.5) \cdot 0.10)) \\ & + (T - 37.5) \\ & \cdot ((T - 37.5) \cdot ((T - 37.5) \cdot 0.0002)) \end{aligned} \quad (3)$$

3.3. Economic analysis

Economic Analysis very crucial determine the effectiveness of the method in large scale and in industrial wastewater system. In industry scale, the wastewater treatment method should be economically viable. In this sub chapter, the economic analysis is made by comparing the total cost needed for EC and integrated EC-membrane system. The total cost needed to be calculated for each method can be divided into 2 main costs which are capital cost and operating cost. There are a few assumptions to be made before proceeding to the calculation of capital cost and operating cost which are the plant is assumed to treat 1000 m³/d of POME with the removal efficiency of 75%. The operating hours per year is assumed to 8000 h per year according to (Sinnott, 2005).

Capital cost is a fixed cost, or a one-time expense incurred on the purchase of an asset such as land, building or equipment needed to produce goods or services. In this study, the capital cost of EC system and membrane system are being calculated to compare to the difference in cost and to assess the cost viability of each treatment method process.

For EC process, the major equipment needed is reactor. The size of reactor can be found by using equation (2) and (3) for EC method and integrated EC-membrane system respectively by determining the removal efficiency as 75% in which are achievable by both EC method and integrated system. The voltage calculated will be 2 V, 4 V, 6 V, 8 V and 10 V. The cost of the reactor can be calculated by using the formula proposed by (Smith, 2005a) which is stated as equation (4) where C_E is equipment cost (\$). C_B is base cost (\$), Q is design capacity (m³), Q_B is base size, M is cost exponent, f_m is correction factor for material, f_p is pressure correction factor, f_t is temperature correction factor and all these values are referred from (Smith, 2005b). The correct factor for EC reactor is set to 1 as the temperature is within 0–50 °C, pressure is at 1 bar while the material selection for the reactor and tank is carbon steel.

$$C_E = C_B \left(\frac{Q}{Q_B} \right)^M f_m f_p f_t \quad (4)$$

After calculating the cost of equipment, the total capital cost can be calculated by multiplying the factors included in the capital

cost estimation. The factors value are obtained by (Smith, 2005a) and Table 1 shows the factors included for physical plant cost calculation which are the equipment erection or installation cost, piping system, instrumentation system and electrical system while total fixed capital cost factors included design and engineering fees, contractor's fees and contingency cost. Equation 5 shows the physical plant cost calculation while Equation 6 shows the total fixed capital cost calculation.

$$\begin{aligned} PhysicalPlantCost(PPC) = & EquipmentCost(C_E) * (1 + f_1 + f_2 \\ & + f_3 + f_4) \end{aligned} \quad (5)$$

$$FixedCapitalCost = PPC * (1 + f_5 + f_6 + f_7) \quad (6)$$

For the total capital cost of integrated EC-membrane system, as there is not much literature information about this method, therefore, it is assumed the plant is having the combination of an EC reactor as well as a membrane tank. The method to calculate the capital cost of EC reactor is similar and the same method is to be used to calculate the capital cost of a membrane tank. Integrated membrane-EC process starts off with a reactor and then to a membrane tank for further treatment.

Operating cost can be defined as the cost of resources to run a specific operation and just to maintain its existence of a business. Operating cost can be divided into 2 sections which are fixed cost and variable cost. Fixed cost is the compulsory cost which needed to run the operating such as raw material cost and utilities cost while variable cost is the cost which may increase or decrease depends on the amount of production such as maintenance.

In this study, the operating cost of EC system was calculated according to (Ghosh, Medhi, & Purkait, 2011). Equation 7 is used to calculate the amount of cost needed per year (MYR/ year) to run the EC operation. In the equation, C_{energy} is the energy (kWh) needed to run the reactor per year while C_{electrode} is the amount of electrode (kg) needed per year for the EC process. Meanwhile, a is price electricity constant for RM0.434/kWh (Malaysia electricity prices, 2019) and b is the price of Fe electrode which is RM2.56/kg (Nshimiyimana, 2020). Membrane cleaning and replacement on the other hand was already included in capital cost under the contractor's fee for the membrane tank.

$$OperatingCost = aC_{energy} + bC_{electrode} \quad (7)$$

The formula to calculate C_{energy} is mentioned in Equation 8 where U is voltage (V), I is current (A), t_{EC} is time for EC (hour), v is flowrate (m³) and C_{electrode} can be calculated via Equation 9 where t is time for EC (second), M_w is molar mass of Fe (56 g/mol), z if electron transfer of Fe (z = 2), and F is faraday's constant (96487C/mol).

$$C_{energy} = \frac{U * I * t_{EC}}{v} \quad (8)$$

$$C_{electrode} = \frac{I * t * M_w}{z * F * v} \quad (9)$$

For the operating cost of integrated membrane EC system, it is calculated as the combination of EC reactor as well as operating cost of a membrane tank. A membrane tank is chosen based on the configuration of the current proposed system that is being upscaled. Besides, the simplicity of the design is one of the advantage and can best be explained by (Lo, McAdam, & Judd, 2018). The operating cost for membrane tank can be calculated as equation 10 according to (Lo et al., 2018) where L_E is price electricity constant for RM0.434/kWh ([Malaysia electricity prices](#), 2019), E_{tot} is total amount of energy needed, L_M is membrane price for RM127.8/m² while M_{tot} is membrane needed per year. Equation 11 shows the membrane size calculation according to (Samhaber & Nguyen, 2014).

$$\text{OperatingCost} = L_E E_{tot} + L_M M_{tot} \quad (10)$$

$$\text{MembraneSize}(m^2) = \frac{\text{VolumeofWastewater}(\frac{L}{d})}{\text{Specificpermeatecapacity}(\frac{L}{m^2d})} \quad (11)$$

3.4. Comparison of cost

Based on the calculation done, the capital cost and operating cost of EC and integrated EC-membrane method are calculated and shown as [Table 4](#) and [Table 5](#) respectively. Based on the table, the lowest capital cost is the EC method with 8 V with only RM 0.58 million needed. This is because the lower the reaction time needed for EC, the smaller the volume of the reactor size and hence, less cost is needed to construct the treatment plant. EC method at 2 V was rendered unsolvable by the modelling due to its result in color removal that did not achieve 75%. This contributes to incapacity in calculating the reactor sizing, capital cost and operating cost subsequently. However, at 4 V, EC method was found to take the longest retention time (60 min) to reach 75% color removal, hence the massive capital cost due to large reactor size. Integrated EC-membrane systems on the other hand tend to have a higher capital cost due to additional membrane tank needed, as well as purchase of membrane sheet. Moreover, based on [Table 5](#), integrated EC-membrane system has the lowest operating cost per year which is only RM 0.13 million with 1 V and RM 0.14 million for 2 V. Integrated EC-membrane method uses lesser operating cost as it lowers the voltage needed to reach the same efficiency compared to EC method and thus, reduce the utilities cost and Fe electrode cost needed per year.

[Table 6](#) shows the total cost that could be calculated in Equation 12 for EC and integrated EC-membrane treatment methods at varying voltages. Based on the results, it can be shown that for EC method, it is best operating the plant with 6 V voltage while integrated EC-membrane method at 2 V to obtain the lowest total cost per year. Integrated EC-membrane at 1 V exhibits a higher capital cost due to longer retention time as compared to 2 V despite the operating cost is lower. When capital cost and operating cost are added up, 2 V shows a lower total cost as compared to 1 V. Therefore, in later calculations, 2 V is considered to represents the inte-

grated EC-membrane system and are to be compared with 6 V for EC method. In addition, from the table, it can be shown that integrated EC-membrane method can achieve the same percentage of removal efficiency by using the lesser amount of energy used.

$$\begin{aligned} \text{TotalCost} = & \text{TotalCapitalCost(CAPEX)} \\ & + \text{TotalOperatingCost(OPEX)} \end{aligned} \quad (12)$$

4 different parameters were compared from this study which were 4 V and 6 V from EC method while for integrated system, both 1 V and 2 V were chosen. The integrated system's color removal efficiency was comparable to EC at 4 V. Nonetheless, as the modelling were done, the time it takes for 4 V method to reach 75% color removal was very high in which causing a larger EC tank. This consequently led to high capital cost as well as high operating cost for EC treatment. Thus, 6 V was chosen to be compared along with 4 V due to its lowest total cost exhibit in [Table 6](#). Despite EC at 6 V have the longest retention time when compared to 8 V and 10 V,

6 V however consumed lesser energy and was reflected by lower operating cost. Integrated system requires a higher capital cost compared to EC. This is because the integrated membrane-EC method requires more equipment setup as a system and hence, increases the capital cost needed. However, the operating cost for integrated system is lower than the EC method. This can be seen as a cost saving method at the long run of industry. Integrated system at 1 V have a higher capital cost due to high retention time as compared to 2 V. Capital cost exhibit by integrated system at 1 V exceeded 2 V as much as RM 146,125. Still, integrated system at 1 V have a lower operating cost due to lower energy consumption.

[Table 6](#) also shows the comparison of total cost for EC and integrated EC-membrane treatment methods looking at a longer period, which is 15 years' time. The table shows that at since year 1, the total cost of the integrated EC-membrane systems (1 V and 2 V) were lower compared to the EC method (4 V and 6 V) and in 15 years, integrated method can save MYR 0.55 million and MYR 0.59 million for 1 V and 2 V respectively compared to the EC method at 6 V. This indicates that the proposed integrated EC-membrane systems are economically feasible and showed a good potential for industrial application in a long run.

Proposed integrated EC-membrane system was proven to be more feasible when compared to EC standalone for treatment of POME. Thus, comparison with conventional biological method is seemed appropriate. A research conducted by Yeoh et al (2005) on anaerobic digestion of POME treatment with an additional heat and power generation from biogas produced (Yeoh, 2005). [Table 7](#) compares the operating condition of integrated EC-membrane with EC and conventional biological method, biological method with gas recovery and coagulation method. The integrated EC-membrane treatment capacity as well as retention time shows better results in the treatment of POME as compared to the other methods mentioned. In addition, reactor volume that exhibit by the proposed EC-membrane system (2 V) also was the smallest in which help in saving space and capital cost. For coagulation and flocculation, the reactor was still in laboratory phase and done in batch system.

Table 4

Total capital cost between EC method and integrated EC-membrane system for 75% colour removal, 1000 m³/d of feed and 8000 h operating hours per year.

Types of Treatment	Retention time (min)	EC Capital Cost (MYR)	Membrane Capital Cost (MYR)	Total Capital Cost (MYR)
EC (2 V)	Unsolvable	Unsolvable	Unsolvable	Unsolvable
EC (4 V)	60.00	913,254	N/A	913,254
EC (6 V)	24.90	614,790	N/A	614,790
EC (8 V)	21.53	575,840	N/A	575,840
EC (10 V)	24.45	609,729	N/A	609,729
Integrated (1 V)	17.40	523,155	272,110	795,264
Integrated (2 V)	11.37	431,975	217,163	649,139

*Unsolvable = method is unable to achieve 75% efficiency, modelling will return an error during calculation of retention time.

Table 5

Total operating cost for EC method and integrated EC-membrane system for 75% colour removal, 1000 m³/d of feed and 8000 h operating hours per year.

Types of Treatment	Utilities Cost (MYR/year)	Fe Electrode Cost (MYR/year)	Membrane Cost (MYR/year)	Total Operating Cost (MYR/year)
EC (2 V)	Unsolvable	Unsolvable	Unsolvable	Unsolvable
EC (4 V)	274,867	4,065	N/A	278,932
EC (6 V)	180,122	4,279	N/A	184,401
EC (8 V)	269,950	5,563	N/A	275,513
EC (10 V)	648,429	9,414	N/A	657,843
Integrated (1 V)	2,097	428	133,125	135,650
Integrated (2 V)	8,222	1,284	133,125	142,631

Table 6

Comparison of Total Cost in 15 years period.

Year	Total Cost (MYR)			
	EC (4 V)	EC (6 V)	Integrated system (1 V)	Integrated system (2 V)
1	1,192,186	799,191	930,915	791,769
2	1,471,118	983,592	1,066,565	934,400
3	1,750,050	1,167,993	1,202,215	1,077,031
4	2,028,981	1,352,395	1,337,865	1,219,661
5	2,307,913	1,536,796	1,473,515	1,362,292
6	2,586,845	1,721,197	1,609,165	1,504,922
7	2,865,777	1,905,598	1,744,815	1,647,553
8	3,144,709	2,090,000	1,880,465	1,790,184
9	3,423,640	2,274,401	2,016,116	1,932,814
10	3,702,572	2,458,802	2,151,766	2,075,445
11	3,981,504	2,643,203	2,287,416	2,218,076
12	4,260,436	2,827,604	2,423,066	2,360,706
13	4,539,368	3,012,006	2,558,716	2,503,337
14	4,818,299	3,196,407	2,694,366	2,645,968
15	5,097,231	3,380,808	2,830,016	2,788,598

Table 7

Comparison of proposed integrated EC-membrane system with EC standalone and conventional methods.

Method	Effluent flowrate	Reaction time	Colour removal	Plant Operating cost	References
EC	1000 m ³ /d	25 min	75%	MYR – 3,380,808	This study
EC-membrane	1000 m ³ /d	11 min	75%	MYR – 2,788,598	This study
Biological treatment	450 m ³ /d	7 days	–	MYR – 8,468,490	(Yeoh, 2005)
Biological treatment with biogas recovery	600 m ³ /h	50 days	–	MYR + 11,638,375	(Lok et al., 2020)
Coagulation and flocculation	1 L (batch)	4 h	95%	–	(Lanan, Selvarajoo, Sethu, & Arumugasamy, 2021)

However, it is worth noting that coagulation and flocculation manage to remove 95% of color from POME with 4 h of settling time. The same trend was usually observed for biological method as the removal of color is relatively high but with also high retention time. In terms of cost, proposed integrated EC-membrane (2 V) has shown the lowest total cost of MYR 2,788,598, followed by EC (6 V) with MYR 3,380,808 and biological method with MYR 8,468,490 over the span of 15 years. However, (Lok, Chan, & Foo, 2020) in their study manage to produce a positive revenue for the plant given the ability to recover valuable by-products such as biogas for resell purposes. This can be done in future studies given the potential for H₂ and O₂ gas recovery from EC process.

4. Conclusions

In conclusion, the experimental data shows that the color removal efficiency of POME by EC method increase with time and with the increasing of voltage. By comparing the experimental result, integrated EC-membrane shows excellent comparable results at 1 V and 2 V with EC standalone method at voltage

higher than 4 V. 75% removal of color for 1 V and 2 V are achievable when using integrated system whereby in EC method, at least 4 V and 60 min are required to achieve such removal efficiency. For model development, both model for EC and integrated EC-membrane method shows a good fit and able to be used for further efficiency calculation. Each model shows an adjusted RSq of higher than 0.8 and a p-value of lower than 0.05 indicates the model fits well. The economic analysis performed shows that integrated EC-membrane method at 2 V has slightly higher capital cost needed but much lower operating cost per year compared to EC method 6 V. Low voltage supplied in integrated system helps the operating cost in terms of electrode and energy consumption remain below EC method even at the first year of implementation. The total cost calculated in 15 years period has observed a possible reduction cost as much as 21.24% (RM 592,210) from EC method. Comparison with conventional biological method with biogas recovery has proven that integrated EC-membrane possess lower operating and capital cost. The production of valuable gaseous such as hydrogen and oxygen gas during EC process could be one of the interesting profitable sources that should be address in future prospect. It can be said

that the proposed method has a good potential in industrial application especially treatment of POME.

Declaration of Competing Interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

Acknowledgement

The authors would like to thank the Researchers Supporting Project No. (RSP-2021/138) King Saudi University, Riyadh, Saudi Arabia.

References

- Abd Halim, N.S., Wirzal, M.D.H., Bilad, M.R., Md Nordin, N.A.H., Adi Putra, Z., Sambudi, N.S., Mohd Yusoff, A.R., 2019. Improving performance of electrospun nylon 6, 6 nanofiber membrane for produced water filtration via solvent vapor treatment. *Polymers* 11 (12), 2117.
- Abdulsalam, M., Che Man, H., Isma Idris, A., Faezah Yunos, K., Zainal Abidin, Z., 2018. Treatment of palm oil mill effluent using membrane bioreactor: novel processes and their major drawbacks. *Water* 10 (9), 1165.
- Abdurakhman, Y., Putra, Z., Bilad, M., Nordin, N., Wirzal, M., Muraza, O. J. A. J. F. S., & Engineering. (2018). Producing biodiesel from waste cooking oil with catalytic membrane reactor: process design and sensitivity analysis. 43(11), 6261–6269.
- Abdurakhman, Y. B., Putra, Z. A., Bilad, M. R., Nordin, N. A. H. M., Wirzal, M. D. H. J. C. E. R., & Design. (2018). Techno-economic analysis of biodiesel production process from waste cooking oil using catalytic membrane reactor and realistic feed composition. 134, 564–574.
- Ali Huddin, I., Irvan, D., Mohd Nordin, A., Arezoo Fereidonian, D., 2012. Comparative study on characterization of Malaysian palm oil mill effluent. *Res. J. Chem. Sci.* 2 (12), 11–15.
- Aris, N.S.M., Ibrahim, S., Arifin, B., Hawari, Y., 2017. Effect of operating parameters on decolorisation of palm oil mill effluent (POME) using electrocoagulation process. *Pertanika J. Sci. Technol.* 25, 197–206.
- Barrera-Díaz, C., Frontana-Uribe, B., Bilyeu, B., 2014. Removal of organic pollutants in industrial wastewater with an integrated system of copper electrocoagulation and electrogenerated H₂O₂. *Chemosphere* 105, 160–164. <https://doi.org/10.1016/j.chemosphere.2014.01.026>.
- Bashir, M.J., Mau Han, T., Jun Wei, L., Choon Aun, N., Abu Amr, S.S., 2016. Polishing of treated palm oil mill effluent (POME) from ponding system by electrocoagulation process. *Water Sci. Technol.* 73 (11), 2704–2712. <https://doi.org/10.2166/wst.2016.123>.
- Bashir, M.J.K., Lim, J.H., Abu Amr, S.S., Wong, L.P., Sim, Y.L., 2019. Post treatment of palm oil mill effluent using electro-coagulation-peroxidation (ECP) technique. *J. Cleaner Product.* 208, 716–727. <https://doi.org/10.1016/j.jclepro.2018.10.073>.
- Bilad, M.R., Azizo, A.S., Wirzal, M.D.H., Jia Jia, L., Putra, Z.A., Nordin, N.A.H.M., Mavukkandy, M.O., Jasni, M.J.F., Yusoff, A.R.M., 2018. Tackling membrane fouling in microalgae filtration using nylon 6,6 nanofiber membrane. *J. Environ. Manage.* 223, 23–28.
- Chafi, M., Gourich, B., Essadki, A.H., Vial, C., Fabregat, A., 2011. Comparison of electrocoagulation using iron and aluminium electrodes with chemical coagulation for the removal of a highly soluble acid dye. *Desalination* 281, 285–292. <https://doi.org/10.1016/j.desal.2011.08.004>.
- Chairunnisaik, A., Arifin, B., Sofyan, H., Lubis, M.R., Darmadi., 2018. Comparative study on the removal of COD from POME by electrocoagulation and electro-Fenton methods: Process optimization. *IOP Conf. Series: Mater. Sci. Eng.* 334. <https://doi.org/10.1088/1757-899x/334/1/012026>.
- Chen, L., Wu, Y., Dong, H., Meng, M., Li, C., Yan, Y., . . . Technology, P. (2018). An overview on membrane strategies for rare earths extraction and separation. 197, 70–85.
- Choong, Y. Y., Chou, K. W., Norli, I. J. R., & Reviews, S. E. (2018). Strategies for improving biogas production of palm oil mill effluent (POME) anaerobic digestion: A critical review. 82, 2993–3006.
- Eliseus, A., Putra, Z., Bilad, M., Nordin, N., Wirzal, M., Jaafar, J., & Khan, A. L. J. A. R. (2018). Energy minimization of a tilted panel filtration system for microalgae filtration: Performance modeling and optimization. 34, 104–115.
- Elma, M., Rahma, A., Pratiwi, A.E., Rampun, E.L.A., 2020. Coagulation as pretreatment for membrane-based wetland saline water desalination. *Asia-Pac. J. Chem. Eng.* 15 (4).
- Ghosh, D., Medhi, C.R., Purkait, M.K., 2011. Techno-economic analysis for the electrocoagulation of fluoride-contaminated drinking water. *Toxicol. Environ. Chem.* 93 (3), 424–437. <https://doi.org/10.1080/02772248.2010.542158>.
- Gong, C., Huang, H., Qian, Y., Zhang, Z., Wu, H., 2017. Integrated electrocoagulation and membrane filtration for PAH removal from realistic industrial wastewater: effectiveness and mechanisms. *RSC Adv.* 7 (83), 52366–52374.
- Hakizimana, J.N., Gourich, B., Chafi, M., Stiriba, Y., Vial, C., Drogui, P., Naja, J., 2017. Electrocoagulation process in water treatment: A review of electrocoagulation modeling approaches. *Desalination* 404, 1–21. <https://doi.org/10.1016/j.desal.2016.10.011>.
- Huang, J., Purvins, A., 2020. Validation of a Europe-wide electricity system model for techno-economic analysis. *Internat. J. Electr. Power Energy Syst.* 123.
- Ibrahim, S., Aris, N.S.M., Ariffin, B., Hawari, Y., Hanafiah, M.A.K.M., 2018. Application of electrocoagulation process for decolorisation of palm oil mill effluent (POME). *Nat. Environ. Pollut. Technol.* 17 (4), 1267–1271.
- Irki, S., Ghernaout, D., Naceur, M.W., Alghamdi, A., Aichouni, M., 2018. Decolorizing methyl orange by fe-electrocoagulation process—A mechanistic insight. *Internat. J. Environ. Chem.* 2 (1), 18.
- Jasni, M.J.F., Arulkumar, M., Sathishkumar, P., Yusoff, A.R.M., Buang, N.A., Gu, F.L., 2017. Electrospun nylon 6, 6 membrane as a reusable nano-adsorbent for bisphenol A removal: Adsorption performance and mechanism. *J. Colloid Interface Sci.* 508, 591–602.
- Lanan, F.A.B.M., Selvarajoo, A., Sethu, V., Arumugasamy, S.K., 2021. Utilisation of natural plant-based fenugreek (*Trigonella foenum-graecum*) coagulant and okra (*Abelmoschus esculentus*) flocculant for palm oil mill effluent (POME) treatment. *J. Environ. Chem. Eng.* 9 (1).
- Latif Ahmad, A., Ismail, S., Bhatia, S., 2003. Water recycling from palm oil mill effluent (POME) using membrane technology. *Desalination* 157 (1), 87–95. [https://doi.org/10.1016/S0011-9164\(03\)00387-4](https://doi.org/10.1016/S0011-9164(03)00387-4).
- Lee, M. D., Osman, M. S., Wong, G. K., & Ngu, H. J. Chemical Oxygen Demand Reduction in Palm Oil Mill Effluent Treatment with Chitosan and Ferric Chloride.
- Lee, M.D., San Lee, P., 2020. Performance of chitosan as natural coagulant in oil palm mill effluent treatment. *Wastewater Treatment*. IntechOpen.
- Lo, C. H., McAdam, E., & Judd, S. (2018). *The cost of a small MBR, WST-EM 15827 1 The cost of a small membrane bioreactor*.
- Lok, X., Chan, Y.J., Foo, D.C.Y., 2020. Simulation and optimisation of full-scale palm oil mill effluent (POME) treatment plant with biogas production. *J. Water Process Eng.* 38.
- Ma, J., Ping, D., & Dong, X. J. M. (2017). Recent developments of graphene oxide-based membranes: a review. 7(3), 52.
- Malaysia electricity prices. (2019). Retrieved from https://www.globalpetrolprices.com/Malaysia/electricity_prices/#:~:text=Malaysia%2C%20December%202019%3A%20The%20price.of%20power%2C%20distribution%20and%20taxes.
- Mat Nawi, N.I., Abd Halim, N.S., Lee, L.C., Wirzal, M.D.H., Bilad, M.R., Nordin, N.A.H., Putra, Z.A., 2020. Improved nylon 6, 6 nanofiber membrane in a tilted panel filtration system for fouling control in microalgae harvesting. *Polymers* 12 (2), 252.
- Medri, W., Costa, R.H.R.D., 2003. A model for cost estimation of system for piggery wastes treatment ponds: a case study. *Pesquisa Operacional* 23, 463–473.
- Nasrullah, M., Singh, L., Mohamad, Z., Norsita, S., Krishnan, S., Wahida, N., Zularisam, A.W., 2017. Treatment of palm oil mill effluent by electrocoagulation with presence of hydrogen peroxide as oxidizing agent and polyaluminum chloride as coagulant-aid. *Water Resour. Indust.* 17, 7–10. <https://doi.org/10.1016/j.wri.2016.11.001>.
- Nshiyimana, J.P., 2020. Techno-economic analysis of electrocoagulation on water reclamation and bacterial/viral indicator reductions of a high-strength organic wastewater—anaerobic digestion effluent. *Sustainability* 12 (7), 2697.
- Saad, M.S., Balasubramanian, L., Wirzal, M.D.H., Abd Halim, N.S., Bilad, M.R., Md Nordin, N.A.H., Adi Putra, Z., Ramli, F.N., 2020a. Integrated membrane-electrocoagulation system for removal of celestine blue dyes in wastewater. *Membranes (Basel)* 10 (8), 184.
- Saad, M.S., Wirzal, M.D.H., Abd Halim, N.S., Khan, M.R., 2020b. Removal color from Palm Oil Mill Effluent (POME): Electrocoagulation Method vs Microfiltration using Nanofiber Membrane. *Internat. J. Electrochem. Sci.* 15, 11283–11293. <https://doi.org/10.20964/2020.11.08>.
- Samhaber, W.M., Nguyen, M.T., 2014. Applicability and costs of nanofiltration in combination with photocatalysis for the treatment of dye house effluents. *Beilstein J. Nanotechnol* 5, 476–484. <https://doi.org/10.3762/bjnano.5.55>.
- Sani, K., Kongjan, P., Pakhathirathien, C., Cheirsilp, B., Sompong, O., Raketh, M., . . . Jariyaboon, R. J. o. W. P. E. (2020). Effectiveness of using two-stage anaerobic digestion to recover bio-energy from high strength palm oil mill effluents with simultaneous treatment. 101661.
- Sinnott, R. K. (2005). *Chemical Engineering Design* (4 Ed.): ELSEVIER BUTTERWORTH HEINAMANN.
- Smith, R., 2005a. *Chemical process design and integration*. Centre for Process Integration, School of Chemical Engineering and Analytical Science. John Wiley & Sons Ltd., University of Manchester.
- Smith, R., 2005b. *Chemical Process: Design And Integration*. John Wiley & Sons.
- Syaichurrozi, I., Sarto, S., Sediawan, W.B., Hidayat, M.J.J., o., W., P., E., 2020. Mechanistic models of electrocoagulation kinetics of pollutant removal in vinashe waste. *Effect Vol.* 36, 101312.
- Tan, Y. D., & Lim, J. S. (2019). Feasibility of palm oil mill effluent elimination towards sustainable Malaysian palm oil industry. 111, 507–522.
- Wastewater Technology Fact Sheet Anaerobic Lagoons. (2002). In: United States Environmental Protection Agency.
- Wirzal, M. D. H., Sathishkumar, P., Alshahrani, L. A., Yusoff, A. R. M., Gu, F. L., Qureshi, M. S., . . . Khokhar, F. M. J. C. P. (2020). Nifedipine degradation by an electro-oxidation process using titanium-based RuO₂–IrO₂–TiO₂ mixed metal oxide electrode. 1–10.

- Wirzal, M. D. H., Yusoff, A. R. M., Zima, J., & Barek, J. J. I. J. o. E. S. (2013). Degradation of ampicillin and penicillin G using anodic oxidation. 8, 8978-8988.
- Xing, J., Liang, H., Chuah, C. J., Bao, Y., Luo, X., Wang, T., . . . Snyder, S. A. J. W. r. (2019). Insight into Fe (II)/UV/chlorine pretreatment for reducing ultrafiltration (UF) membrane fouling: Effects of different natural organic fractions and comparison with coagulation. 167, 115112.
- Yavuz, Y., Ogunveren, U.B., 2018. Treatment of industrial estate wastewater by the application of electrocoagulation process using iron electrodes. *J. Environ. Manage.* 207, 151–158. <https://doi.org/10.1016/j.jenvman.2017.11.034>.
- Yeoh, B. J. I. E. J. (2005). A technical and economic analysis of heat and power generation from biomethanation of palm oil mill effluent. 6,
- Zahari, A.M., Shuo, C.W., Sathishkumar, P., Yusoff, A.R.M., Gu, F.L., Buang, N.A., Lau, W.-J., Gohari, R.J., Yusop, Z., 2018. A reusable electrospun PVDF-PVP-MnO₂ nanocomposite membrane for bisphenol A removal from drinking water. *J. Environ. Chem. Eng.* 6 (5), 5801–5811.
- Zainal, N.H., 2018. A review on the development of palm oil mill effluent (pome) final discharge polishing treatments. *J. Oil Palm Res.* 29 (4), 528–540. <https://doi.org/10.21894/jopr.2017.00012>.
- Zhang, M., Yao, L., Maleki, E., Liao, B.-Q., Lin, H., 2019. Membrane technologies for microalgal cultivation and dewatering: Recent progress and challenges. *Algal Res.* 44,. <https://doi.org/10.1016/j.algal.2019.101686> 101686.