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Markers of Anthropogenic Contamination: A Validated Method for Quantification of Pharmaceuticals, Illicit Drug Metabolites, Perfluorinated Compounds and Plasticisers in Sewage Treatment Effluent and Rain Runoff

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Abstract

An effective, specific and accurate method is presented for the quantification of 13 markers of anthropogenic contaminants in water using solid phase extraction (SPE) followed by high performance liquid chromatography (HPLC) tandem mass spectrometry (MS/MS). Validation was conducted according to the International Conference on Harmonisation (ICH) guidelines. Method recoveries ranged from 77-114% and limits of quantification between 0.75-4.91 ng/L. A study was undertaken to quantify the concentrations and loadings of the selected contaminants in 6 sewage treatment works (STW) effluent discharges as well as concentrations in 5 rain-driven street runoffs and field drainages. Detection frequencies in STW effluent ranged from 25% (ethinylestradiol) to 100% (benzoylecgonine, bisphenol-A (BPA), bisphenol-S (BPS) and diclofenac). Average concentrations of detected compounds in STW effluents ranged from 3.62 ng/L (ethinylestradiol) to 210 ng/L (BPA). Levels of perfluorinated compounds (PFCs) perfluorooctanoic acid (PFOA) and perfluorononanoic acid (PFNA) as well as the plasticiser BPA were found in street runoff at maximum levels of 1160 ng/L, 647 ng/L and 2405 ng/L respectively (8.52, 3.09 and 2.7 times more concentrated than maximum levels in STW effluents respectively). Rain-driven street runoff may have an effect on levels of PFCs and plasticisers in receiving rivers and should be further investigated. Together, this method with the 13 selected contaminants enables the quantification of various markers of anthropogenic pollutants: inter alia pharmaceuticals, illicit drugs and their metabolites from humans and improper disposal of drugs, while the plasticisers and perfluorinated compounds may also indicate contamination from industrial and transport activity (street runoff).

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1. Introduction

From the turn of the 21st century, the occurrence, fate and environmental toxicity of pharmaceuticals and other so-called emerging contaminants in the aquatic environment (rivers, reservoirs, oceans and drinking water) has been the subject of many studies (e.g., Kolpin et al., 2002; Vajda et al., 2008; 2011; Llorca et al. 2012) and reviews (e.g., Petrie et al., 2015; Wilkinson et al., 2015). Nearly every human and veterinary pharmaceutical has been detected in the aquatic environment, typically at sub-ug/L concentrations. Evidence suggests that once in the aquatic environment, residues of some pharmaceuticals and other emerging contaminants (ECs) cause biological disruption/dysfunction in exposed non-target organisms via mechanisms such as endocrine dysfunction (Woodling et al., 2006; Vajda et al., 2008; Patisaul et al., 2009; Vajda et al., 2011). Once administered, pharmaceuticals are rarely, if ever, completely metabolised in the body and are directed to sewage treatment works (STWs) via urinary and biliary fecal excretions (Jjemba, 2006; Silva et al., 2011). Residues of pharmaceuticals typically enter the aquatic environment through STW effluent outfalls where many compounds do not completely degrade (Verlicchi et al., 2012). Other anthropogenic contaminants such as musk compounds used in fragrances and lotions, surfactants, and ultraviolet filters in sunscreen are also directed to STWs via wastewater (Peck, 2006; Roosens et al., 2007). Supplementary Table 1 shows concentrations of 21 pharmaceuticals and other emerging contaminants which are typically detected in river water.

Emerging contaminants are commonly defined as any compound (both synthetic and natural) in the environment whose presence is not routinely monitored and shows the potential to cause ecological disruption (Raghav et al., 2013; United States Geological Survey, 2014). Among such ECs are perfluorinated compounds (PFCs) such as perfluorocotanoic acid (PFOA), perfluorononanoic acid (PFNA), perfluorobutane sulfonic acid (PFBS), and perfluorocotane sulfonate (PFOS) as well as plasticisers including bisphenol-A (BPA) and bisphenol-S (BPS) among others. Plasticisers are commonly used to modify the rigidity of plastic materials and are used in product and food packaging, manufacture of polycarbonate plastic, thermal receipt paper, water pipes, epoxy resins, vinyl floors, car tires, among many others (U.S. Food and Drug Administration, 2015). Perfluorinated compounds are commonly used in waterproofing agents, polishes, waxes, non-stick cookware and cleaning products (Posner, 2012). ECs such as plasticisers and PFCs are also known to enter the aquatic environment through STW effluent outfalls (Meyer and Bester, 2004; Zhang et al., 2013).

Biodegradation of PFC precursor molecules during sewage treatment into other PFCs is suggested to increase their effluent concentrations by 32-290 times over that in the influent (Zhang et al., 2013). For example, Zhang et al. (2013) reported that of 28 sampled STWs in economically developed areas of China, effluent concentrations of PFOA were higher than influent 81% of the time, 69% of the time for perfluorononanoic acid (PFNA), 57% of the time for perfluorobutane sulfonic acid (PFBS) and 58% of the time for PFOS. Precursor molecules found in STW influent include N-ethyl perfluorooctane sulphonamide, which degrades into PFOS and fluorotelomer alcohols (e.g., 8:2 fluorotelomer alcohol) which degrade into PFOA (Zhang et al., 2012; Avendaño and Liu, 2015).

Recent evidence indicates that PFCs may be introduced into the aquatic environment through non-point sources, such as runoff from streets (Zushi et al., 2008; Murakami et al., 2009; Furl et al., 2011). Zushi et al. (2008) reported concentrations of PFCs 2-11 times higher in rain runoff than in the effluent outfalls of three studied STWs along the Tsurumi River (Yokohama City, Kanagawa Prefecture, Japan). Despite evidence of PFCs introduced to the aquatic environment through runoff, there are limited studies quantifying similar contributions of other ECs, such as plasticisers. Furthermore, studies comparing STW effluent to runoff contributions of PFCs and plasticisers into the same river system is limited.

Here, a validated method is presented to accurately and specifically quantify 13 markers of anthropogenic contamination in fresh water: *inter alia* pharmaceuticals, illicit drugs and their metabolites. Pharmaceuticals and illicit drugs are specific indicators of human-derived contamination and improper disposal of drugs, while the plasticisers and perfluorinated compounds also indicate contamination from industrial and transport activity (street runoff). The presented method can be used to efficiently and specifically quantify concentrations of contaminants of human, industrial and transportation origins. A key feature of this method is the achievement of adequate detection limits using a relatively small amount of water (200mL) for analysis. Here, increased ease of sample transport (due to the small volume of water required) enables analysis of a greater number of samples in relatively little time. The sensitivity/detection limits of the method proposed as such can be further increased by increasing the amount of sample water used in solid phase extraction (SPE). The developed method enabled a comparison between contaminant inputs into selected rivers via sewage treatment effluent and runoff from both streets and fields.

2. Materials and Methods

2.1. Standards and Reagents

Thirteen compounds were selected for analysis including pharmaceuticals: ethinylestradiol, acetaminophen and diclofenac, illicit drugs and metabolites: methamphetamine and its metabolite amphetamine and benzoylecgonine (metabolite of cocaine), plasticisers: bisphenol-S (BPS), bisphenol-A (BPA) and its biotransformation product 4'-hydroxyacetophenone (HAP), perfluorinated compounds: PFOA, PFOS, PFNA, and PFBS, and deuterated internal standards: BPA-D16, acetominophen-D4 and methamphetamine-D5. All compounds were purchased from Sigma Aldrich (Gillingham, Dorset, U.K.) and were of 96% purity or higher. PFOS was purchased as a potassium salt and benzoylecgonine was tetrahydrated. Strata-X 33µm polymeric reversed phase 200 mg/6mL SPE cartridges were purchased from Phenomenex (Macclesfield, Cheshire, U.K.). A Phenomenex Kinetex 2.6u C18 150x2.1mm chromatography column using a SecurityGuard ULTRA UHPLC C18 2.1mm guard column was purchased from Phenomenex (Macclesfield, Cheshire, U.K.). Whatman GF/F-grade glass microfiber filters (diameter 47mm, pore size 0.7 μm) were purchased from Fisher Scientific (Loughborough, Leicestershire, U.K.).

Stock solutions were prepared for each compound and internal standards at a concentration of 1000 mg/L in LCMS-grade acetonitrile except for PFOS, diclofenac and ethinylestradiol, which were prepared at 1000 mg/L in HPLC grade acetone. A mixed solution was made in LCMS-grade acetonitrile to a concentration of 5 mg/L for all analytes except BPA at 30 mg/L, ethinylestradiol at 25 mg/L, and HAP at 7.5 mg/L which was further diluted into working solutions. Concentrations of target analytes in the mixed solution were optimised based on spectrometer sensitivity to each respective compound. Solutions were stored in 10mL borosilicate glass volumetric flasks at 4°C in the dark and renewed monthly. A mixture of all compounds at 100 ng/L was analysed via HPLC-MS/MS once per week for 4-weeks to ensure no significant degradation occurred during monthly use of standard solutions.

2.2. Sample Collection

Effluent samples (n=12) were collected in duplicate from six STWs along three rivers in the south east of England (Hogsmill, Bourne and Blackwater: Supplementary Table 2). Runoff samples were collected from grass field drainage (n=3) and street runoff (n=2) during periods of rainfall (Supplementary Table 3).

Grab samples were collected (200 mL) in amber glass bottles. All bottles were washed three times with 50/50 acetonitrile:acetone (v/v) followed by three washes with milli-Q water and were rinsed with sample water before collection. Water samples were collected from

STW effluent outfalls and drainage pipes (both field and street) directly prior to entering the river flow. After collection, samples were stored at 4°C in the dark until vacuum filtration (within 8 h of collection) using GF/F glass membrane filters. Membrane filters were soaked in 10% nitric acid for 12 hours (to degrade any organic compounds within or on the filter) followed by three rinses under vacuum filtration with 50:50 acetonitrile:acetone (v/v) and three rinses with HPLC water. All samples were extracted before analysis using the SPE procedure described in Section 2.3 within 24 h of collection.

2.3. Solid Phase Extraction

Solid Phase Extraction (SPE) of 200 mL sample water was carried out using Phenomenex Strata-X cartridges preconditioned with 3 mL 50:50 acetonitrile/acetone (v/v), washed with 3 mL HPLC-grade H₂O and loaded at a rate of 5 mL/min. Loaded cartridges were dried under vacuum for 15 min and eluted with 2x7mL aliquots of 50:50 acetonitrile/acetone (v/v) at a rate of 1 mL/min. Extracts were evaporated to dryness using rotary evaporation and reconstituted with 1 mL of (80:20 HPLC H₂O/acetonitrile, v/v) spiked with internal standards to 25 ng/mL.

2.4. HPLC-MS/MS Analysis

Extracts were analysed by HPLC-MS/MS using an Agilent Technologies 1260 Infinity HPLC and an Agilent Technologies 6430 series triple quadrupole mass spectrometer operated in multiple reaction monitoring (MRM) mode. Adequate separation was obtained at a flow rate of 0.2 mL/min with a solvent gradient starting at 80:20 H₂O/acetonitrile (v/v) reducing to 35:65 H₂O/acetonitrile (v/v) at 15 min which returned to 80:20 H₂O/acetonitrile (v/v) at 15.1 min, where it remained until the end of the run at 19.5 min (Figure 1). Collision energy, fragmentor and electron multiplier voltage were optimised for two transition ions per target analyte and internal standard using electrospray ionisation (Table 1). An electron multiplier voltage of +/-350 V was used for every target compound. A capillary voltage of +/-4000 V, chamber current of 0.12 µA, nebuliser pressure of 30 psi, gas temperature of 325°C and gas flow of 12 L/min were used for mass spectrometry. An injection volume of 10 µL was used for all analysis. Quantification was achieved using a 9-point blank-offset and internal standard-corrected calibration (calibration points also underwent SPE) calculated by linear regression analysis in Agilent MassHunter Workstation Software: Quantitative Analysis (version B.04.00/Build 4.0.225.19). Calibration standards and blanks also underwent SPE and were used to normalise for recoveries. The method was validated using the ICH Harmonised Tripartite Guideline for Validation of Analytical Procedures: Q2(R1) Text and Methodology (ICH, 2005).

2.5. Method Validation and Quality Assurance

2.5.1 Specificity

Specificity was assessed by spiking river water collected from the Rowhill Nature Reserve (source of the Blackwater River) with a mixture of all selected PPCPs/ECs at a concentration of 100 ng/L. Spiked sample chromatograms were compared to those of non-spiked water samples and assessed for the presence of interfering peaks at the specific retention time of known target compounds.

2.5.2 Intra-/Inter-day Precision/Repeatability

Repeatability and precision were determined using the relative standard deviation (RSD) calculated from replicate extractions of spiked HPLC water subjected to SPE. Intraday (intermediate) precision was calculated at two concentrations, 25 ng/L (n=3) and 100 ng/L (n=3) every day over 3 days (6 determinations/day). Inter-day precision (repeatability) was calculated at two concentrations, 25 ng/L (n=6) and 100 ng/L (n=6). Each sample was analysed three times by HPLC-MS/MS in triplicate and averages are reported in Table 2.

2.5.3 Accuracy

Method accuracy was evaluated by spiking 200 mL HPLC-grade water with a mix of analytes at three concentrations followed by solid phase extraction and analysis (triplicate analysis, 9 total determinations). Three concentrations of mixed analytes selected were: 10 ng/L (n=3 replicates), 50 ng/L (n=3 replicates) and 100 ng/L (n=3 replicates). Spiked water was subjected to the same treatment as experimental samples (SPE followed by rotary evaporation and reconstitution in 1 mL mobile phase before HPLC-MS/MS analysis). Replicates were offset by a procedural blank of 200 mL HPLC-water (stored in an identical beaker as the accuracy replicates) subjected to SPE, rotary evaporation and reconstitution. The absolute value of the difference between the observed/calculated concentrations and the known spiked concentration are presented as accuracy (Table 3).

2.5.4 Recovery

Recovery was evaluated by spiking 200 mL HPLC-grade water with two concentrations of all (mixed) analytes: 50 ng/L (n=3 replicates) and 100 ng/L (n=3 replicates). Spiked HPLC water was subjected to the same treatment as experimental samples (SPE followed by rotary evaporation and reconstitution in 1 mL mobile phase before HPLC-MS/MS analysis). Samples were analysed in triplicate and analyte concentrations calculated using linear regression with calibration points having also undergone SPE. Calculated analyte

concentrations were divided by the known spiked concentration and multiplied by 100 to yield % recovery (Table 3). HPLC water was selected to evaluate recoveries over spiked matrix (STW effluent) due to the high variability of selected contaminant levels found between studied STW effluent outfalls.

2.5.4. Linearity

Linearity was assessed through calibration curve r^2 -values using linear regression analysis with the aim of attaining r^2 -values >0.95. Analyte-spiked 200 mL aliquots of HPLC water were subjected to SPE followed by rotary evaporation and reconstitution in 1 mL mobile phase before HPLC-MS/MS analysis. The spiked HPLC-water concentrations ranged from 1-500 ng/L for ethinylestradiol, 0.6-300 ng/L for BPA, 0.3-150 ng/L for HAP and 0.2-100 ng/L for all other compounds (Supplementary Table 4).

2.5.5. Limit of detection/quantification

Both the limit of detection (LOD) and limit of quantification (LOQ) were determined using a signal-to-noise approach. A 100 mL aliquot of HPLC-grade water was spiked (pre-SPE) with analyte masses equal to that in the lowest calibration level (BPA 1.2 ng/L, ethinylestradiol 1.0 ng/L, hydroxyacetophenone 0.3 ng/L and all others 0.2 ng/L). SPE followed by rotary evaporation and reconstitution in 1mL 80:20 H₂O/acetonitrile (v/v) spiked with internal standards was conducted prior to HPLC-MS/MS analysis. A total of ten replicates were used for the determination of LOD and LOQ. LOD and LOQ were determined as the compound's concentration that yielded a mean signal-to-noise ratio of 3:1 and 10:1 respectively (n=10). Where such S/N values were not determined at the lowest calibration level, identical replicate analysis was performed at increasing concentrations until adequate S/N ratios were found (Table 3).

2.5.6. Calibration and quantitative analysis

A 9-point internal standard-corrected and blank offset calibration curve was determined for each analyte just before field sample extraction and analysis on every sample collection day. Calibration points were assessed by spiking 200mL aliquots of HPLC-grade water to the concentrations in Supplementary Information Table 4. All calibration and blank samples were subjected to the same storage, solid phase extraction, rotary evaporation and reconstitution methods/conditions as experimental samples explained in sections 2.2-2.4. Using this method, background levels of any potential method-derived contamination was offset in each calibration and further offset by a procedural blank. Linear regression analysis was performed using Agilent MassHunter Workstation Software: Quantitative Analysis (version B.04.00/Build 4.0.225.19). The calibrated range was compound-specific: 1-500 ng/L for ethinylestradiol, 0.6-300 ng/L for BPA, 0.3-150 ng/L for hydroxyacetophenone and 0.2-100 ng/L for all other compounds (Supplementary Table 4).

2.5.7 Quality Assurance and Procedural Blanks

Throughout the development and application of the presented method, great care was given to quality, accuracy, and cleanliness. Extracts of environmental samples were always injected into the LCMS in triplicate. After every six injections a standard quality control SPE-extract (200 mL HPLC-water spiked to 100 ng/L of each analyte) was run to check instrument parameters and ensure accurate quantification. All glassware was, without exception, rinsed three times with a mixture of acetonitrile:acetone (50:50, v/v) between every use except for round bottom flasks used in rotary evaporation which were additionally first scrubbed with water and soaked in 10% nitric acid for 16 h between uses. Volumetric flasks used to store solutions were dedicated to specific compounds or mixtures as to avoid cross-contamination. Laboratory workspaces were cleaned before and after every use and the laboratory itself was thoroughly cleaned on a weekly basis.

Procedural blanks were used throughout method validation and quantitative analysis. Blanks underwent the same sample storage, solid phase extraction, rotary evaporation and reconstitution as all calibration points and field samples. For example, procedural blanks used in calibration and quantitative analysis were created using 200 mL HPLC-water stored in the same 3-times solvent rinsed amber glass bottles as field samples and stored in the dark at 4° C until solid phase extraction followed by rotary evaporation of SPE extract, reconstitution and HPLC-MS/MS analysis. The only difference between these procedural blanks and true field blanks was that they did not undergo the sample collection step (capturing of water from the effluent outfall). Similarly, the 200 mL HPLC-water used for validation blanks were stored in the same solvent rinsed beakers and followed identical handling procedures as respective validation replicates (per standard good laboratory practice). Use of blanks in method validation followed that outlined in the ICH Q2(R1) Text and Methodology (ICH, 2005). Analysis was carried out to determine levels of target analytes in blanks relative to a calibration of standard samples (not undergoing identical SPE, storage and handling procedures). Of the thirteen target contaminants, only two were detected in procedural blanks used for quantitative analysis (n=3). Here, BPA was detected below the LOQ and hydroxyacetophenone was detected at 1.22 ng/L on average. In procedural blanks used for method validation, only hydroxyacetophenone was detected and was consistently found below the LOQ (n=3). It should be noted that all calibration points used for quantitative analysis underwent the same handling procedures and sample treatment as field samples and were blank offset.

3. Results

3.1. Method Validation

Validation results are tabulated in Tables 2 and 3 and a brief description is included in this section.

All intra-day (intermediate precision) RSD-values were below 10% except for methamphetamine at 25 ng/L (10.1% on day 3) and 100 ng/L (12.5% day 1 and 13.7% day 3), acetaminophen at 25 ng/L (12.2% on day 3) and 4'-hydroxyacetophenone at 25 ng/L (10.5% day 2 and 12.3% day 3). Inter-day (repeatability) RSD-values were consistently below 10% except for methamphetamine at 100 ng/L (11.8%). Table 2 shows average repeatability and intra-day precision data.

Method accuracy was determined to be within $\pm 0.03-2.15$ ng/L for all compounds at a concentration of 10 ng/L, $\pm 0.66-11.4$ ng/L at 50 ng/L and $\pm 0.64-19.8$ ng/L for all compounds at a 100 ng/L concentration (Table 3).

Recoveries ranged from 77.6% to 106% at an analyte concentration of 50 ng/L and from 80.1% to 114% at 100 ng/L (Table 3).

Calculated r²-values were consistently above 0.97 indicating a strong linearity for all studied compounds over a range of 1-500 μ g/L for ethinylestradiol, 0.6-300 μ g/L for BPA, 0.3-150 μ g/L for hydroxyacetophenone and 0.2-100 μ g/L for all other compounds (Table 3).

Limits of detection ranged from 0.22 ng/L to 1.17 ng/L (amphetamine and ethinylestradiol respectively) and limits of quantification ranged from 0.75 ng/L to 4.91 ng/L (PFNA and ethinylestradiol respectively). Limits of detection/quantification are shown in Table 3.

3.2. Occurrence of PPCPs/ECs in Sewage Treatment Effluent

PPCPs and ECs were identified in every STW effluent (Table 4). It should be noted that summary statistics were calculated using all values found above the LOQ. Of the thirteen target compounds, eleven were detected in at least one sample, two (amphetamine and methamphetamine) were not detected in any sample. Detection frequencies ranged from 25% (ethinylestradiol) to 100% (benzoylecgonine, BPA, BPS and diclofenac). Table 4 shows concentrations and a basic statistical analysis of PPCPs/ECs detected in STW effluent outfalls. Across all studied compounds, the mean detection frequency was 78.5%. BPA was

found at a maximum level of 890 ng/L in STW 2 effluent which was significantly higher than any other studied compound in STW effluent outfalls (2.72 times higher than the next highest concentration).

Both BPA and its main biodegradation product 4'-hydroxyacetophenone (HAP) were typically found at the highest concentrations of any target contaminant in studied STW effluent outfalls. Except for STW 1, higher HAP/BPA ratios (possibly indicating greater inplant degradation of BPA into HAP) were observed in the effluent outfall of STWs using activated sludge treatment (HAP/BPA ratio 1.28-1.52) than STWs using trickling filter contact process (HAP/BPA ratio 0.14-0.22).

Of the pharmaceutical compounds detected, diclofenac was typically found at the highest concentration with an average of 62.2 ng/L (1.84 times that of the next highest pharmaceutical, acetaminophen). Of the three recreational drugs and urinary metabolites thereof, only benzoylecgonine was detected. Benzoylecgonine, a significant urinary metabolite of cocaine, was found in 100% of effluent samples with an average concentration of 9.20 ng/L. Perfluorinated surfactants were detected with frequencies from 58.3% (PFBS) to 91.7% (PFOA and PFNA). PFBS was detected at the highest average level of all PFCs (51.3 ng/L) while PFNA was detected at the highest maximum level of the PFCs (209 ng/L in STW 4 effluent).

Daily loads of the compounds detected entering respective rivers via STW effluent outfalls (Table 5) ranged from 0.05 g/day (benzoylecgonine in STW 1 first collection) to 22.7 g/day (BPA in STW 2 second collection). On average, BPA showed the highest daily load between all STWs of 6.14 g/day which was 2.68 times that of the next highest, 2.29 g HAP/day (BPA's main biodegradation product). Except for STW 2, total daily loads of discharged PPCPs/ECs seemed to be correlated with the population equivalents of the respective STW facilities (see Supplementary Table 2 for STW characteristics, Table 5 for loads). STWs with higher population equivalents showed greater total daily loads of selected contaminants (except for STW 2).

3.3. Occurrence of PPCPs/ECs in Street and Field Drainage

Seven of the thirteen compounds monitored were identified in the drainage samples, all either plasticisers or PFCs: hydroxyacetophenone, BPS, BPA, PFBS, PFOA, PFNA and PFOS (Table 6). Street runoff into the River Bourne and the Blackwater River (inputs 1 and 2) showed the highest concentrations of target contaminants. Overall, field drainage showed both the lowest levels of detected contaminants and the lowest number of detected compounds/site (average 4.7 compounds/site detected in field drainage vs 6 compounds/site

in street runoff). The highest measured concentration was of BPA (2410 ng/L) in street runoff into the River Bourne (Input 1), 2.7 times more concentrated than the highest detected level in STW effluent. PFOA and PFNA were also identified at maximum levels of 1160 ng/L and 647 ng/L in street runoff into the Blackwater River (Input 2) respectively which were 8.52 (PFOA) and 3.09 (PFNA) times higher than maximum respective levels found in STW effluent outfalls. Of the seven compounds detected in field drainage, average concentrations were from 3.14 times (BPA) to 24.9 times (PFBS) lower than average concentrations in STW effluent.

Compounds of human origin (pharmaceuticals and residues of recreational drugs) were not detected in any field drainage sample. An accurate average discharge volume of each drainage input was not possible to determine thus daily loads of detected contaminants were not evaluated in sampled runoffs.

4. Discussion and Conclusion

The presented SPE HPLC-MS/MS method can effectively and specifically quantify levels of the selected contaminants in water. The method was validated in-house using the ICH Harmonised Tripartite Guideline for Validation of Analytical Procedures: Q2(R1) Text and Methodology (ICH, 2005). The presented method demonstrated that pharmaceuticals are not the only contaminants of ecological concern entering rivers via STW effluent discharge. Specifically, plasticisers and perfluorinated surfactants were detected at similar or higher levels than pharmaceuticals in STW effluent discharges. Plasticisers were found to be ubiquitous contaminants in all examined discharges into rivers (STW, field, and street runoff).

The type of sewage treatment used appears to have an effect on contaminant effluent levels of BPA. Higher ratios of HAP (BPA degradation product) to the BPA parent compound (HAP:BPA ratio) may indicate greater in-plant degradation of BPA into HAP. HAP:BPA ratios in the effluent of STWs using activated sludge ranged from 1.28 to 1.52 and were higher than those observed in plants using trickling filter contact where HAP:BPA ratios ranged from 0.14-0.22 (except for STW 2). This finding corroborates those found by Barber et al. (2012) where improved removal efficiency of endocrine disrupting compounds was observed in a sewage treatment facility in Boulder, Colorado (USA) after an upgrade from trickling filter contact to activated sludge treatment. However, contaminants of ecological concern are not solely introduced into the aquatic environment via STW effluent outfalls.

During periods of rainfall, drainage from streets was found to contain PFCs and plasticisers at maximum levels, significantly higher (2.7 times for BPA, 3.09 times for PFNA and 8.52 times for PFOA) than their maximum levels in STW outfalls. It should be noted that the volume of street runoff (introduced into rivers via point-source drainage pipes) is highly variable and almost, if not always, exclusively less than the volume of a typical STW effluent discharge. Hence, the load of such contaminants entering rivers via street runoff is unknown, but may still be influential when considering the great number of such drainage point sources (albeit with lower relative flow rates) when compared to the number of STW effluent discharges along the course of a river. This is a finding that warrants further investigation.

Drainage from suburban fields into rivers was also found to contain small amounts of plasticisers and PFCs. Average concentrations were 3.14 to 24.9 times lower in field drainage than in STW effluent. Relatively low levels of plasticisers and PFCs entering river flow via field drainage may be explained by the presence of such chemicals in components of pipes used in channelling the discharge into rivers (U.S. Food and Drug Administration, 2015) or via their presence in the atmosphere (Huang et al., 2012; Liu et al., 2015).

Further work is warranted to elucidate the impact of street runoff on levels of plasticisers and perfluorinated compounds in receiving rivers. Here, limited samples were selected solely as reference to other sources of such contaminants and as a brief comparison to levels found in STW effluent discharges. Finding high levels of PFCs and plasticisers in street runoff is significant and should be further investigated. A study comparing first-flush rainfall discharges with time-weighted averages is particularly worth investigating.

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6.0 Tables and Figures

Table 1- Ion Fragmentation and Spectrometer Conditions

Compound	Compound	Precursor	Polarity	Quantifier	Qualifier	Fragmentor	Collision Energy
Туре		Ion Mass		Ion Mass	Ion Mass		(eV)
51		(amu)		(T1)	(T2)		(T1, T2)
Perfluorinated	PFBS	298.8	Negative	80.0	99.0	128	30, 24
Compounds	PFNA	462.9	Negative	419.0	219.0	92	4, 8
	PFOA	412.9	Negative	369.0	169.0	100	4, 8
	PFOS	498.4	Negative	80.0	99.0	180	30, 30
Pharmaceuticals	Acetominophen	152.1	Positive	110.3	65.3	92	12, 30
	Diclofenac	293.9	Negative	250.0	214.0	56	4, 12
	Ethinylestradiol	295.5	Positive	133.0	199.5	92	20, 12
Plasticisers	Bisphenol-A	227.1	Negative	212.1	133.1	116	12, 20
	Bisphenol-S	249.0	Negative	108.1	92.1	132	24, 30
	4'-Hydroxyacetophenone	135.0	Negative	92.1	120.0	100	24, 12
Recreational	Amphetamine	136.1	Positive	119.1	91.0	66	5,17
Drugs	Benzoylecgonine	290.1	Positive	168.3	105.2	92	16, 30
	Methamphetamine	150.1	Positive	91.0	65.1	92	17, 45
Internal	Acetominophen-D4	156.1	Positive	114.4	69.4	100	12, 32
Standards	Bisphenol A-D16	241.1	Negative	141.8	233.0	130	38, 13
	Methamphetamine-D5	155.1	Positive	121.1	91.1	95	54, 17

T1- Transition ion 1 T2- Transition ion 2

Table 2- Intra-/Inter-day Repeatability/Precision

Compound	Intra-day	<u>r RSD (%)</u>	Inter-day R	SD (%)				
		25ng/L			100ng/L	25ng/L	100ng/L	
	Day 1	Day 2	Day 3	Day 1	Day 2	Day 3		
4'-Hydroxyacetophenone	5.92	10.5	12.3	7.47	8.99	4.03	9.57	6.83
Acetaminophen	5.73	7.4	12.4	6.01	3.82	6.48	8.53	5.44
Amphetamine	3.91	7.13	5.48	9.48	9.37	5.44	5.51	8.10
Benzoylecgonine	7.83	10.5	7.99	4.58	3.42	4.14	8.78	4.05
BPA	5.04	6.56	3.37	3.67	3.43	3.91	4.99	3.67
BPS	3.06	4.71	4.00	4.10	1.74	4.34	3.92	3.40
Diclofenac	6.01	6.10	4.75	6.34	4.87	6.66	5.62	5.96
Ethinylestradiol	9.54	8.57	9.92	9.33	8.59	8.89	9.34	8.94
Methamphetamine	6.79	9.93	10.1	12.4	9.28	13.7	8.95	11.8
PFBS	6.65	4.63	9.76	6.62	4.66	7.20	7.01	6.16
PFNA	2.24	3.15	2.62	3.19	1.77	3.28	2.67	2.75
PFOA	3.05	4.37	2.78	3.72	2.07	3.74	3.40	3.17
PFOS	5.64	3.75	6.72	5.11	3.09	2.18	5.37	3.46

Compound	<u>pKa</u>	Accuracy	Accuracy (ng/L)		Recovery (%)		<u>Linearity</u>	Det./Quant. Lim	its	
		10ng/L	50ng/L	100ng/L				LOD	LOQ	
		(+/-)	(+/-)	(+/-)	50ng/L	100ng/L	r ² -value	(ng/L)	(ng/L)	
4'-Hydroxyacetophenone	8.12	0.91	3.96	16.8	92.1	83.2	0.987	0.31	1.04	
Acetaminophen	9.38	1.47	4.9	6.71	90.2	93.3	0.973	0.28	0.93	
Amphetamine	10.1	2.15	9.91	16.6	80.2	83.4	0.969	0.22	1.09	
Benzoylecgonine	3.15-9.54	2.1	11.4	19.8	77.2	80.1	0.997	0.31	1.02	
Bisphenol-A	10.3	0.15	2.89	13.9	105	113	0.994	1.17	3.87	
Bisphenol-S	8.2	2.4	11.2	15.7	77.6	84.3	0.993	0.34	3.87	
Diclofenac	4.15	0.41	0.66	4.43	101	104	0.986	0.29	0.96	
Ethinylestradiol	10.7	1.49	5.16	9.67	89.7	90.3	0.971	0.98	4.91	
Methamphetamine	9.87	0.03	3.47	9.68	93.1	90.3	0.988	0.32	1.06	
Perfluorobutanesulfonic Acid	<1	1.1	1.84	6.86	103	93.1	0.974	0.33	1.13	
Perfluorononanoic Acid	<1	1.13	3.34	0.64	93.3	101	0.993	0.23	0.75	
Perfluorooctanoic Acid	<1	1.04	1.63	2.32	103	97.7	0.997	0.34	1.13	
Perfluorooctane Sulfonate	<1	1.46	4.97	0.27	90.1	99.7	0.989	0.51	1.52	

 Table 3- Method validation results

LOD- Limit of detection, LOQ- Limit of quantification

	Effluent	Effluent	Effluent	Effluent	Effluent	Effluent					
	1	2	3	4	5	6					
Population Equivalent	383,000	86,800	37,000	140,000	34,300	88,400					
Average daily flow (m ³)	58,180	25,534	10,073	31,196	9,098	24,795					
Average flow (m ³ /sec)	0.673	0.296	0.117	0.361	0.105	0.287					
Treatment	AST	AST/TFC*	AST	AST	TFC	TFC					
Compound (ng/L)							Mean (ng/L)	Det. Freq. (%)	Max (ng/L)	SD (ng/L)	RSD (%)
4'-Hydroxyacetophenone	13.1	213	150	144	11.8	58.5	98.4	91.7	327	82.8	84.1
Acetaminophen	17.5	133	17.3	10.3	5.63	19.1	33.8	91.7	157	48.9	145
Amphetamine	ND	ND	ND	ND	ND	ND	ND	0	ND		
Benzoylecgonine	1.09	19.9	10.8	7	4.61	11.9	9.2	100	25.7	6.57	71.4
BPA	216	488	98.5	113	82.1	264	210	100	891	154	73.2
BPS	5.76	201	25	34.2	9.71	4.13	46.5	100	306	76.6	165
Diclofenac	76.2	31.5	102	98.3	38.9	26.2	62.2	100	201	34.2	55
Ethinylestradiol	ND	5.11	<loq< td=""><td>3.75</td><td>ND</td><td>ND</td><td>4.43</td><td>25</td><td>10.2</td><td>3.48</td><td>78.6</td></loq<>	3.75	ND	ND	4.43	25	10.2	3.48	78.6
Methamphetamine	ND	ND	ND	ND	ND	ND	ND	0	ND		
PFBS	9.64	21	57.6	92.3	96	31.3	51.3	58.3	115	36.8	71.7
PFNA	6.03	5.48	91.9	113	14.9	19.4	41.8	91.7	209	47.8	114
PFOA	19.9	22.2	21.1	3.89	36.3	87.1	31.7	91.7	136	28.9	91.4
PFOS	60.1	31.4	22.2	19.8	3.46	56.1	32.2	83.3	115	22.1	68.5
*Combination of Trickling Filter Contact (TFC) and Activated Sludge Treatment (AST) ND- Not Detected											

Table 4- STW information and concentrations (ng/L) of PPCPs/ECs detected in STW effluents (n=12) with basic statistical analysis

*Combination of Trickling Filter Contact (TFC) and Activated Sludge Treatment (AST)

	1						
Compound (g/day)	STW 1	STW 2	STW 3	STW 4	STW 5	STW 6	Mean (g/day)
4'-Hydroxyacetophenone	0.35-1.17	2.51-8.36	0.03-2.99	ND-9.00	0.03-0.19	0.50-2.40	2.29
Acetominophen	0.92-1.11	2.80-4.01	0.11-0.24	0.14-0.50	0.01-0.10	ND-2.95	0.91
Amphetamine	ND	ND	ND	ND	ND	ND	0
Benzoylecgonine	0.05-0.07	0.36-0.66	0.03-0.18	0.14-0.30	0.02-0.06	0.18-0.41	0.21
BPA	1.09-24.1	2.17-22.7	0.25-1.74	2.50-4.55	0.20-1.29	5.98-7.09	6.14
BPS	0.13-0.54	2.46-7.82	0.02-0.49	0.06-2.07	0.01-0.16	0.06-0.14	1.16
Diclofenac	4.02-4.85	0.09-1.52	0.03-2.02	0.11-6.02	0.14-0.67	0.40-1.30	1.72
Ethinylestradiol	ND	ND-0.29	ND	ND-0.23	ND	ND	0.04
Methamphetamine	ND	ND	ND	ND	ND	ND	0
PFBS	0.26-0.86	ND-1.07	ND-1.16	ND-2.88	ND-0.87	ND-1.50	1.03
PFNA	0.16-0.55	0.07-0.21	0.29-1.56	0.52-6.53	ND-0.14	0.40-0.56	0.93
PFOA	0.53-1.78	0.26-0.87	0.13-0.29	ND-0.24	0.19-0.47	0.94-3.37	0.76
PFOS	0.29-6.70	0.10-1.51	ND-0.45	ND-0.62	ND-0.03	0.20-2.59	1.09
Mean sum of all (g/day)	24.8	29.9	6.01	19.9	2.76	14.3	16.3

 Table 5- Range of compound loads (g/day) entering receiving rivers via respective STW effluent outfalls (n=12)

Compound	Street Runot	ff (ng/L)		Field Draina	ge (ng/L)
	Input 1	Input 2	Input 3	Input 4	Input 5
4'-Hydroxyacetophenone	48.4	254	5.98	16.5	16.7
Acetominophen	ND	ND	ND	ND	ND
Amphetamine	ND	ND	ND	ND	ND
Benzoylecgonine	ND	ND	ND	ND	ND
Bisphenol-A	2410	511	54.2	45.6	101
Bisphenol-S	50.2	40.6	2.27	4.42	159
Diclofenac	ND	ND	ND	ND	ND
Ethinylestradiol	ND	ND	ND	ND	ND
Methamphetamine	ND	ND	ND	ND	ND
PFBS	79.1	ND	ND	ND	6.19
PFNA	69.2	648	ND	23.0	2.01
PFOA	6.52	1160	ND	15.5	ND
PFOS	8.97	ND	ND	ND	6.57

Table 6- Concentrations of PPCP/ECs detected in street runoff and field drainage prior to entering rivers



*Quantifying transition ion **Figure 1-** Chromatogram of target PPCP/ECs